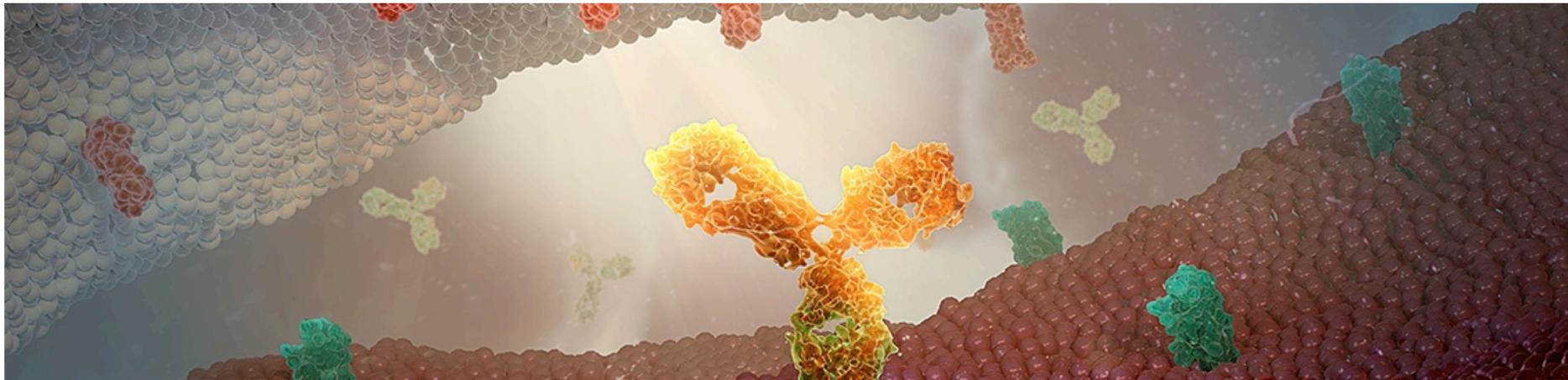


The use of automation and machine learning in Flow Chemistry

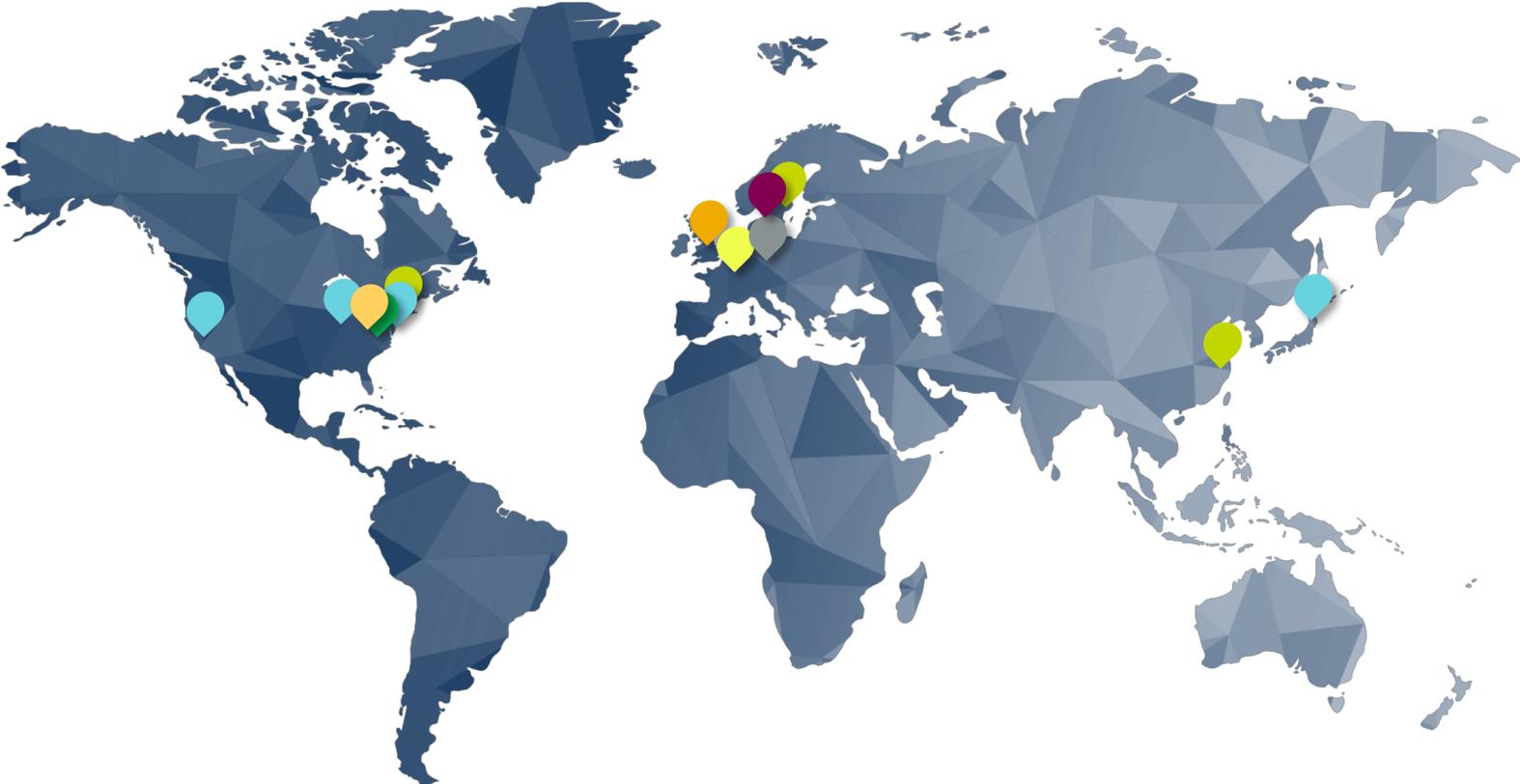
Dr Graeme Clemens

RSC Process Chemistry and Technology Group Launch Symposium

19 November 2018



AstraZeneca



AstraZeneca UK



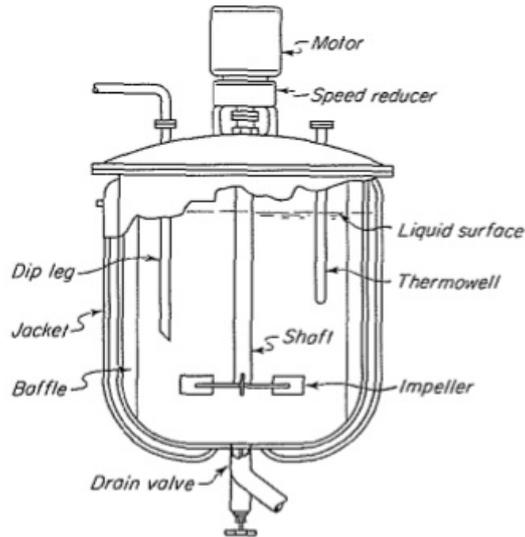
AstraZeneca:

- Creates around £4.5 billion in GVA directly
- Generates direct exports of £7.05 billion representing about 3% of the UK total exports of goods
- Invests £1.3 billion in R&D in the UK annually
- Headquarters located in Cambridge



Batch Chemistry

Batch chemistry mainly employed at AstraZeneca



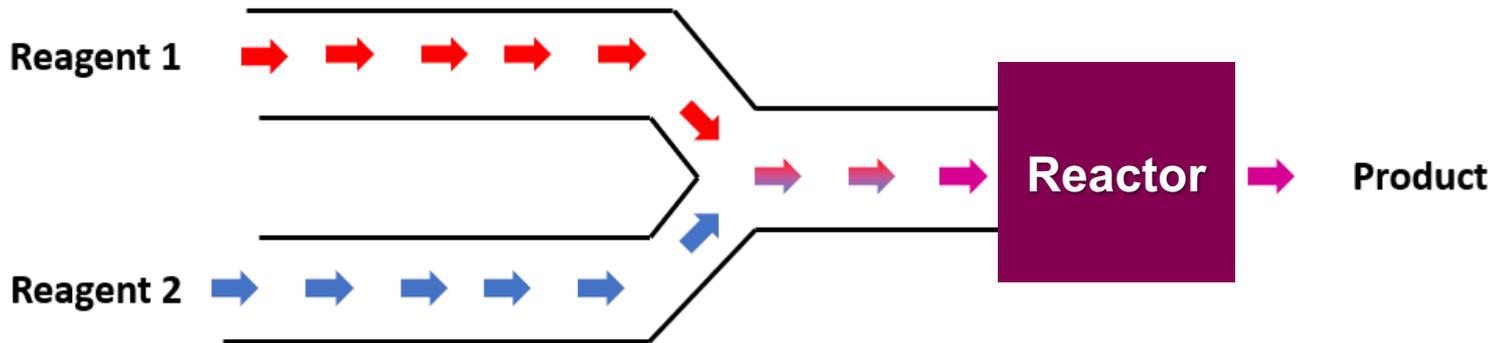
- Thousands of litres of material required when manufacturing on a large scale.
- Some reactions are too dangerous to be considered in batch.
- When things go wrong there is a significant material loss and financial cost.
- Batch chemistry scale up from the laboratory to manufacturing scale can be challenging.



Flow Chemistry

Reagents are continuously pumped through the reactor

Product is continuously collected from end of reactor

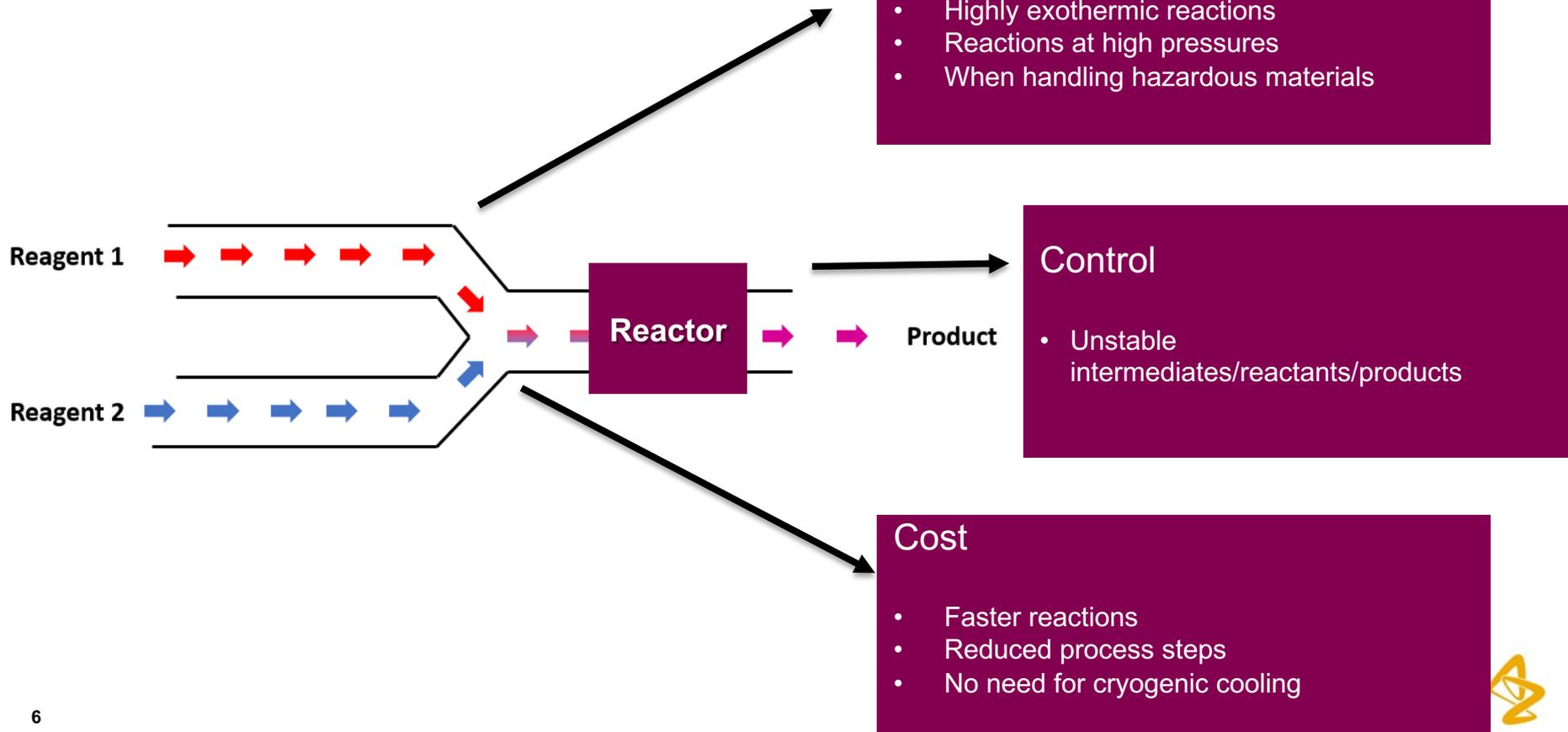


Benefits

- Better heat transfer (larger surface area)
- Much better reaction control than Batch chemistry
- Easier to scale-up than batch chemistry
- Lower inventories (safer)
- Easier to respond to product being out of specification
- Ease of Automation when compared to batch chemistry



AZ Drivers for Flow Chemistry



Targeted Continuous Processing Platforms

PRIMARY PLATFORMS

Cryogenic



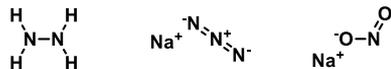
[M] = Li, Mg

S_NAr



Nu = NHR, NR¹R²

Hazardous



$\text{Na}^+ \text{C}\equiv\text{N}^-$

NO_2^+

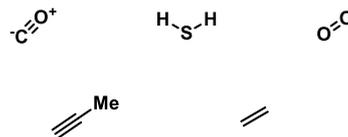
High Temp



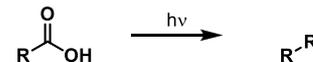
Boc deprotections, Cyclisations,
Decarboxylations etc.

DEVELOPING PLATFORMS

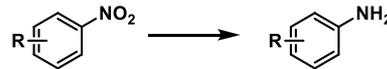
Gas/Liquid



Photoredox

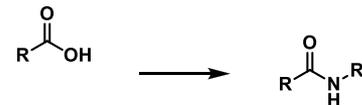


Solid Support

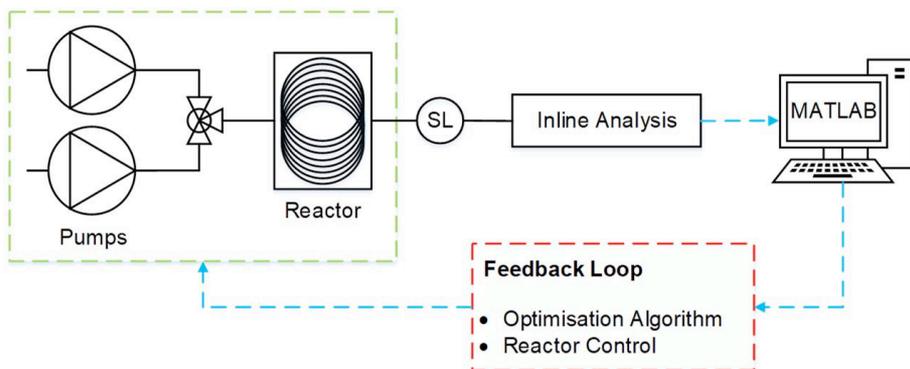


Pd/C reductions, debenzylations

Amide formation

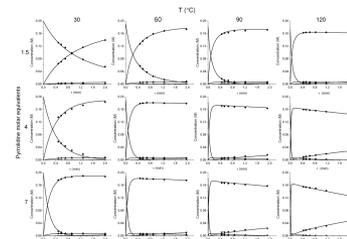


Self-Optimising Flow Reactor (SOFR)



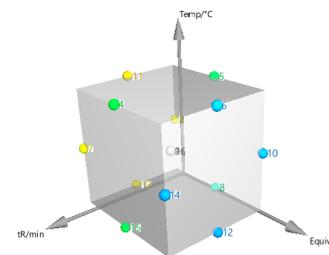
Reaction Kinetics

- Fit rate constants, orders and activation energies



Design of Experiments (DOE)

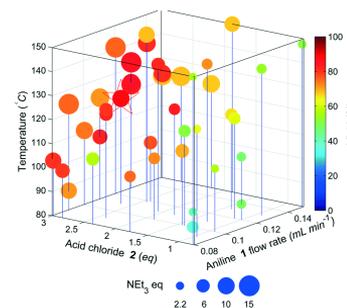
- Process optimisation and understanding



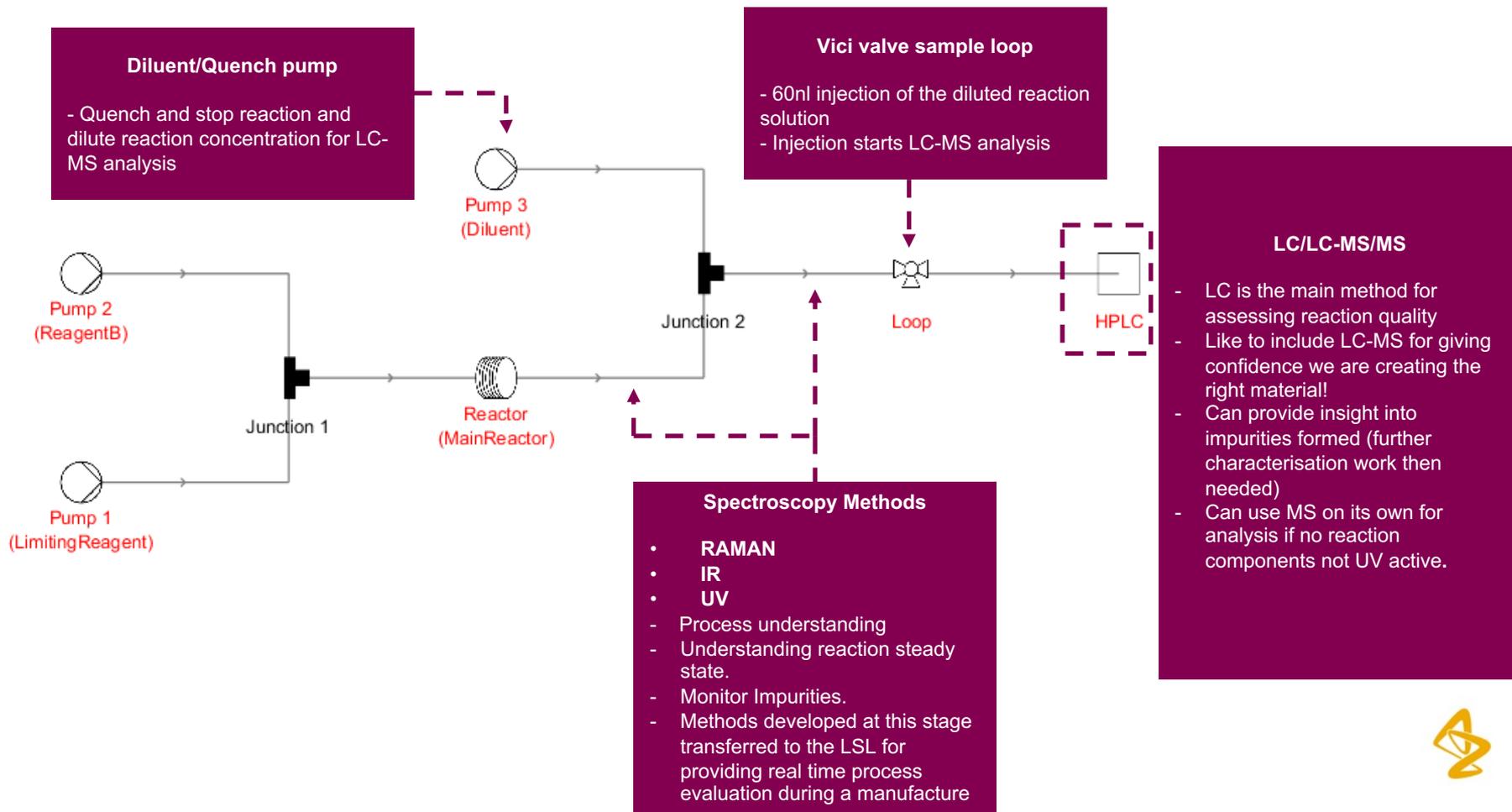
Self-optimisation (AI)

- Self-learning algorithm uses inline analytics to assess reaction space to search for a true optimum

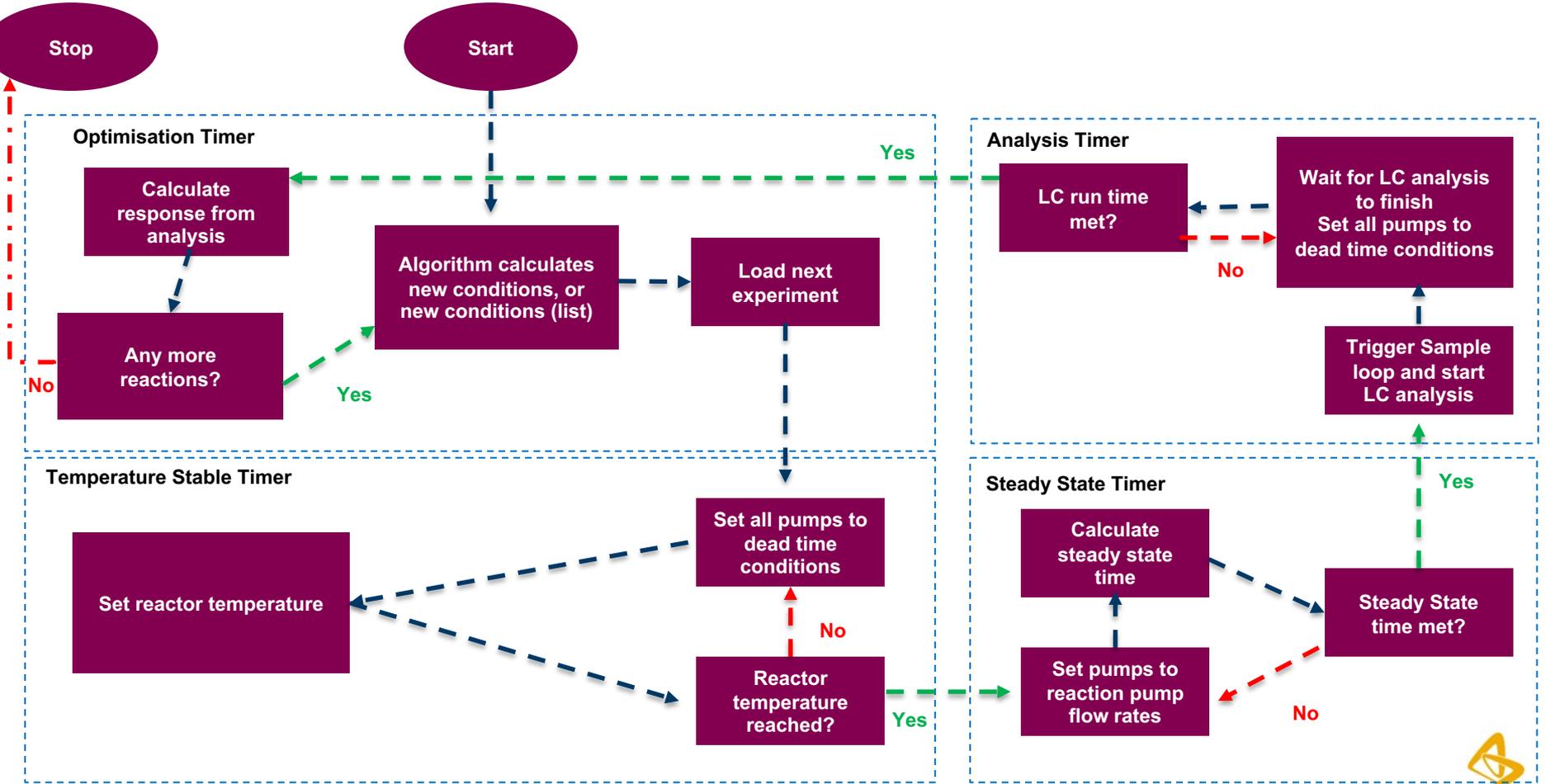
- More likely to find a true optima than DOE
- Quick reaction optimisation with minimal effort



Analytical Techniques for Reaction Assessment



SOFR System Flow



Advantages of Automation

Efficient Use of Time

| | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 |
|----|--------|----------|-----------|-----------|-------------|-------------------|----------------|------|
| 1 | Exp No | Exp Name | Run Order | Incl/Excl | temperature | molar equivalence | residence time | HPLC |
| 2 | 5 | N5 | 1 | Incl | 125 | 4.5 | 5 | |
| 3 | 8 | N8 | 2 | Incl | 170 | 4.5 | 11 | |
| 4 | 10 | N10 | 3 | Incl | 147.5 | 5.5 | 5 | |
| 5 | 13 | N13 | 4 | Incl | 147.5 | 4.5 | 8 | |
| 6 | 1 | N1 | 5 | Incl | 125 | 3.5 | 8 | |
| 7 | 6 | N6 | 6 | Incl | 170 | 4.5 | 5 | |
| 8 | 11 | N11 | 7 | Incl | 147.5 | 3.5 | 11 | |
| 9 | 3 | N3 | 8 | Incl | 125 | 5.5 | 8 | |
| 10 | 2 | N2 | 9 | Incl | 170 | 3.5 | 8 | |
| 11 | 15 | N15 | 10 | Incl | 147.5 | 4.5 | 8 | |
| 12 | 4 | N4 | 11 | Incl | 170 | 5.5 | 8 | |
| 13 | 7 | N7 | 12 | Incl | 125 | 4.5 | 11 | |
| 14 | 12 | N12 | 13 | Incl | 147.5 | 5.5 | 11 | |
| 15 | 14 | N14 | 14 | Incl | 147.5 | 4.5 | 8 | |
| 16 | 9 | N9 | 15 | Incl | 147.5 | 3.5 | 5 | |

- System can perform a list of reactions sequentially with in-line analysis overnight.
- Analyse analytical data in the morning.

Automation of DOE design

- 16 reactions with a residence time range of 5-11 minutes performed overnight.
- Flow chemist would need 2 days in the lab to do the same.

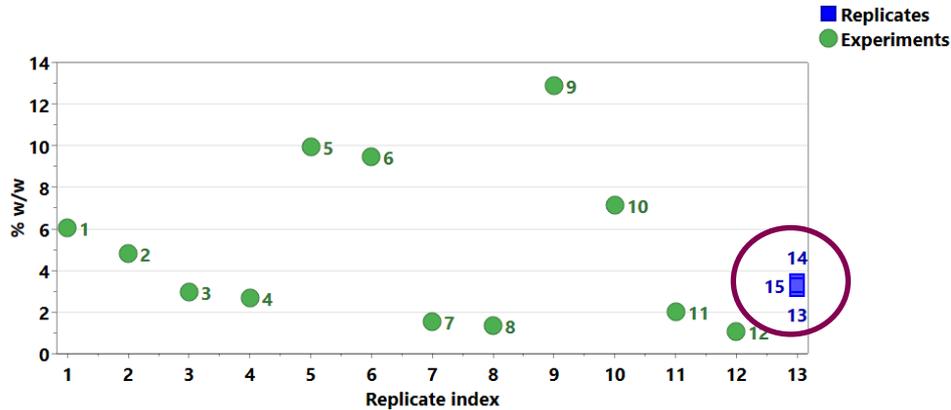
Machine Learning Optimisation

- 80 reactions performed over ~35 hours.
- Fully autonomous self-learning algorithm.
- No human intervention.

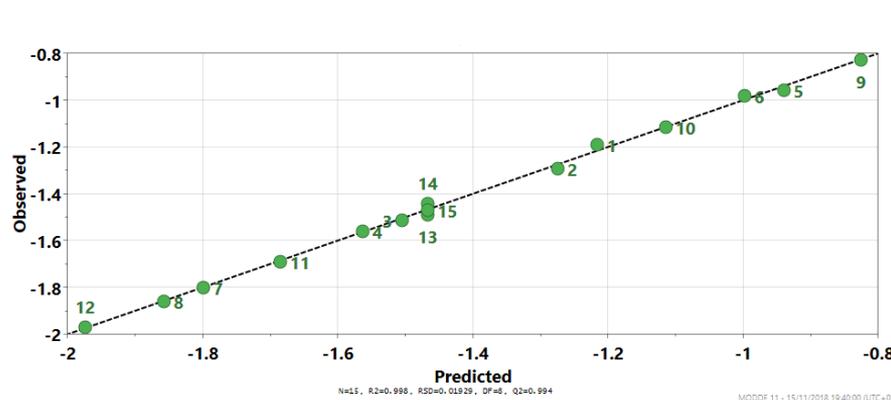


Advantages of Automation

Quality of Analytical data



- Excellent repeatability.
- Good quality MLR models.
- Increased confidence in predicting the process and understanding important process parameters.

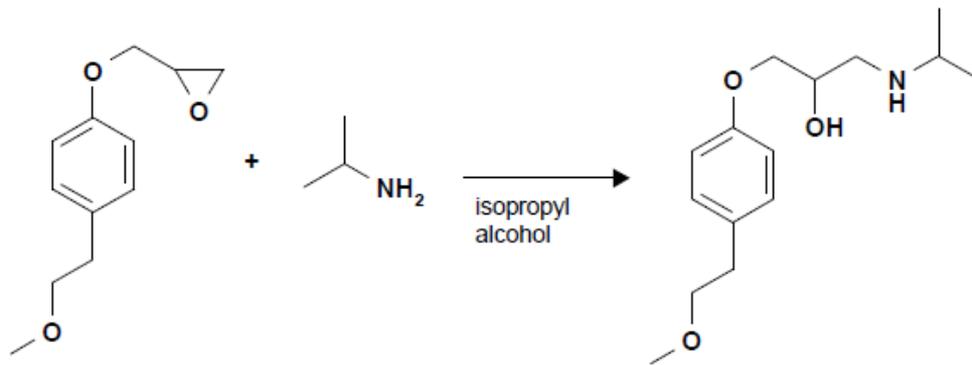


Removal of human error!



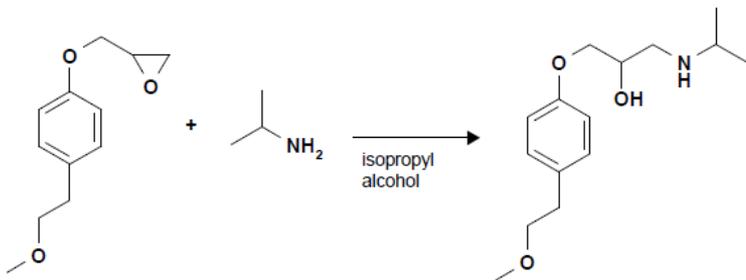
SOFR Case Study 1- Transfer of Batch Reaction to Flow

Developing a Flow Process as an Alternative to Aminolysis Batch Reaction



SOFR Case Study 1- Transfer of Batch Reaction to Flow

Nucleophilic epoxide ring opening reaction



Batch reaction

- Temperature: 60C
- Duration: 4 hours
- 3 molar equivalence of isopropyl amine

- All-in batch reaction - not inherently safe unless run in a pressure rated batch reactor
- Continuous processing could be an alternative to make the reaction inherently safe (flow reactors have superior pressure rating).

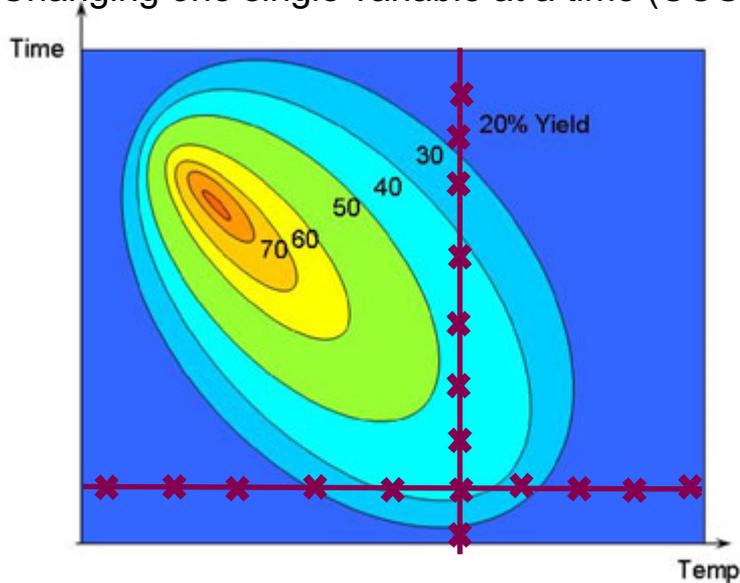
Aims

1. Develop flow process using SOFR instrument.
2. Demonstrate the potential of this type of automated flow reactor platform for automating structured statistical designs (design of experiments (DOE)).
3. Find key parameters of the process for driving reaction conversion and maximising product yield.

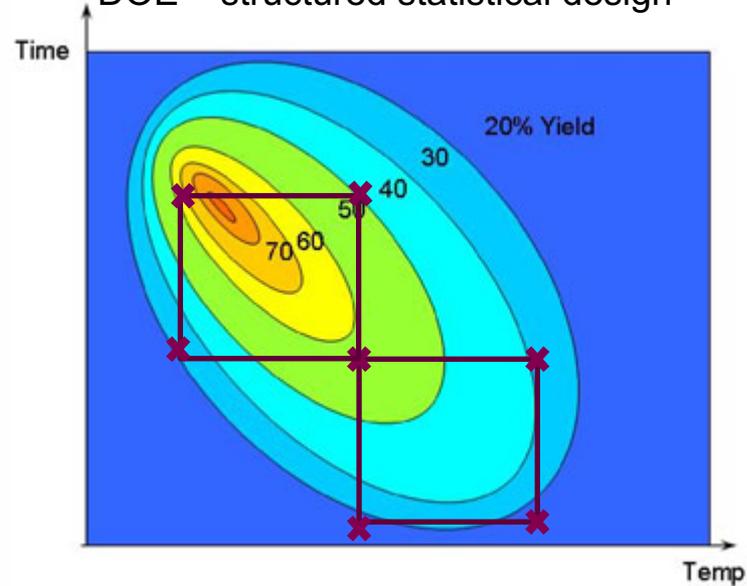


Design of Experiments (DOE)

Changing one single variable at a time (COST)



DOE – structured statistical design



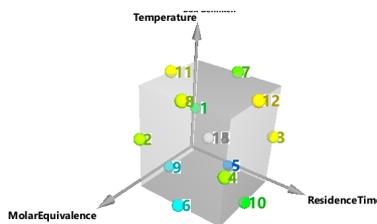
DOE – an efficient way (reduced experimentation) of understanding important parameters and their effect on a process.



SOFR Case Study 1- Transfer of Batch Reaction to Flow

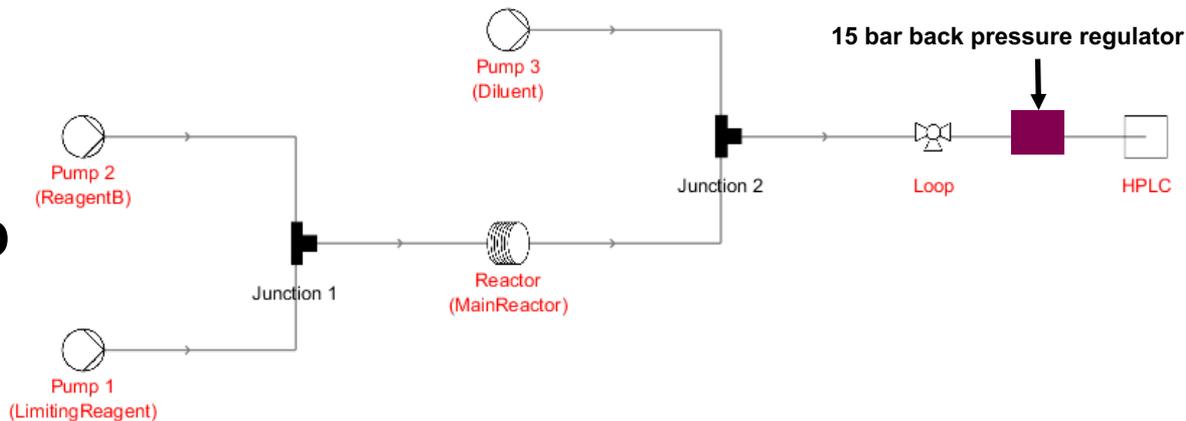
| DOE Factors | Ranges |
|--------------------------------------|-----------|
| Temperature | 125 - 170 |
| Molar equivalence of Isopropyl amine | 3.5 – 5.5 |
| Residence time (minutes) | 5 - 10 |

Box Behnken DOE Design (MODDE)



- 16 reactions performed overnight
- Design has 3 factor levels – linear and quadratic responses
- Minimal confounding
- LC-MS used as analytical technique

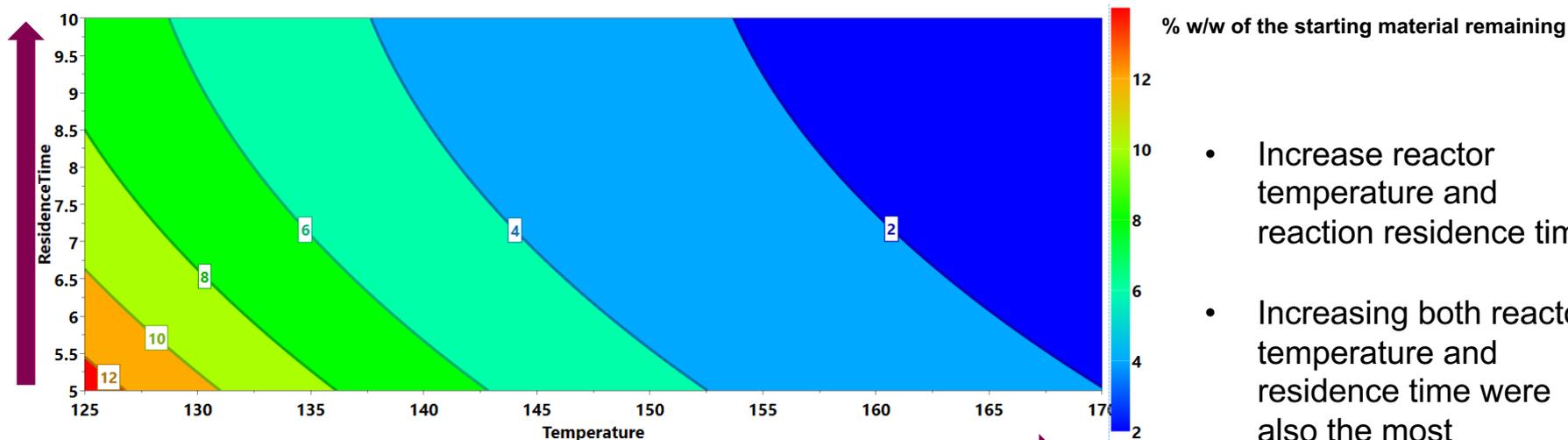
SOFR setup



SOFR Case Study 1- Transfer of Batch Reaction to Flow

Results

Process parameters driving conversion of limiting reagent?

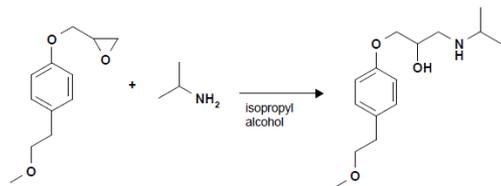


- Increase reactor temperature and reaction residence time
- Increasing both reactor temperature and residence time were also the most significant factors for product yield

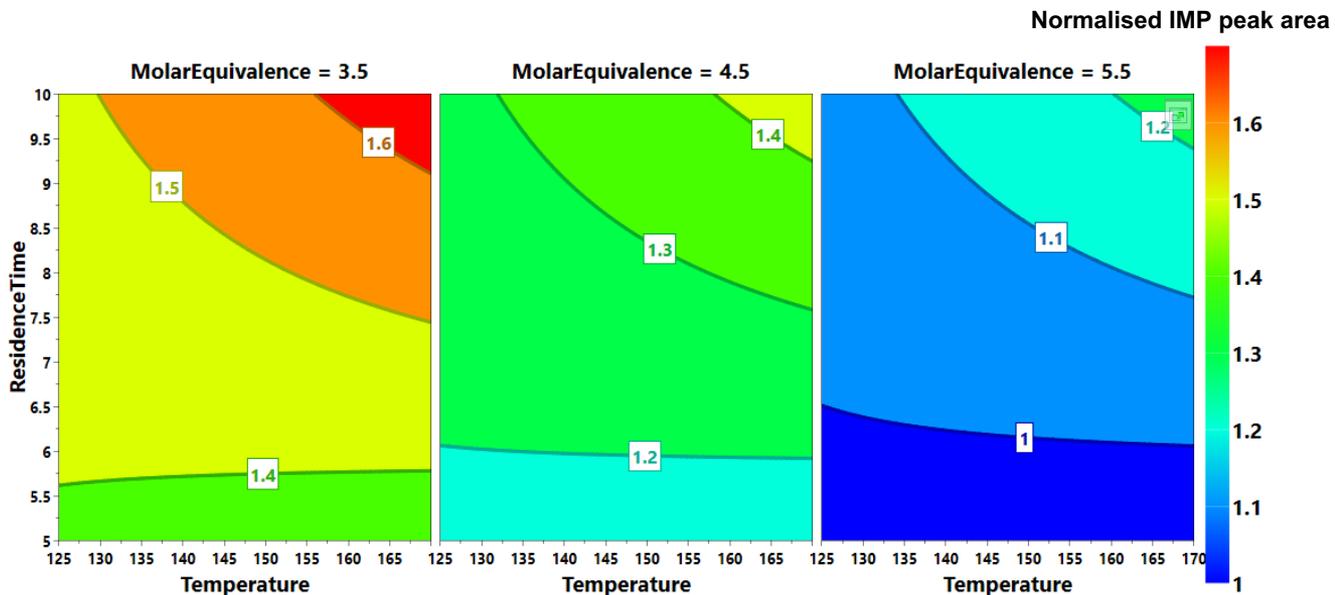
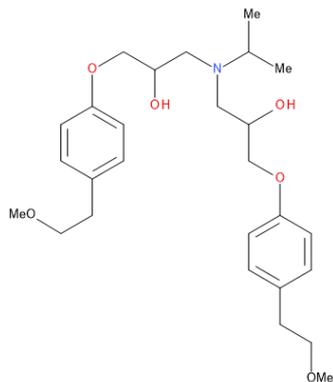


SOFR Case Study 1- Transfer of Batch Reaction to Flow

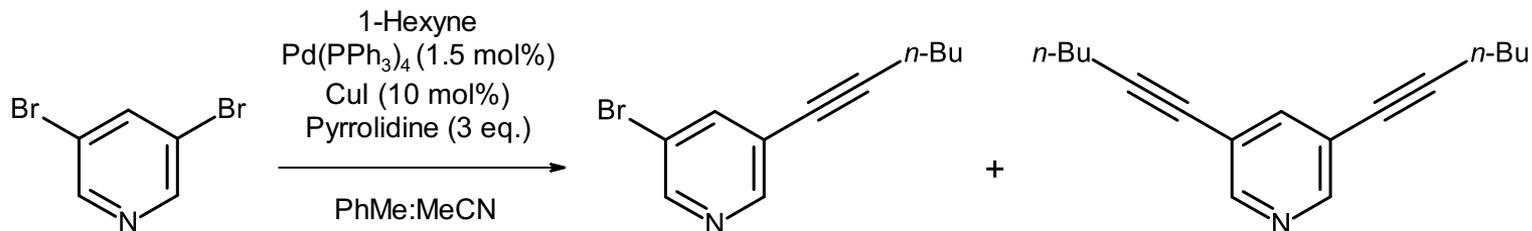
Process parameters important for the formation of the main impurity?



Main impurity



Optimise Sonogashira Coupling Reaction using the PARETO Multi-Objective Self- Learning Algorithm

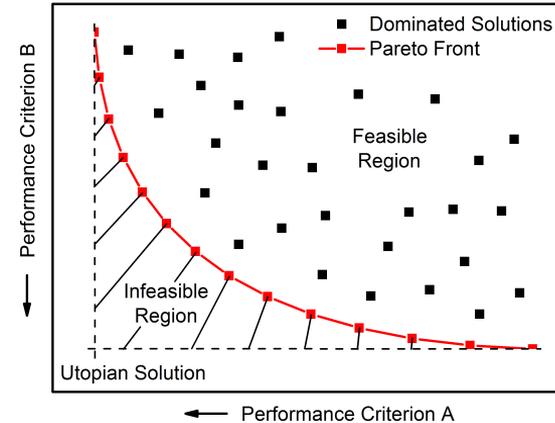


SOFR Case Study 2- Machine Learning for Process Optimisation

PARETO Multi-Objective Algorithm

Typically more than 1 objective when optimising a process

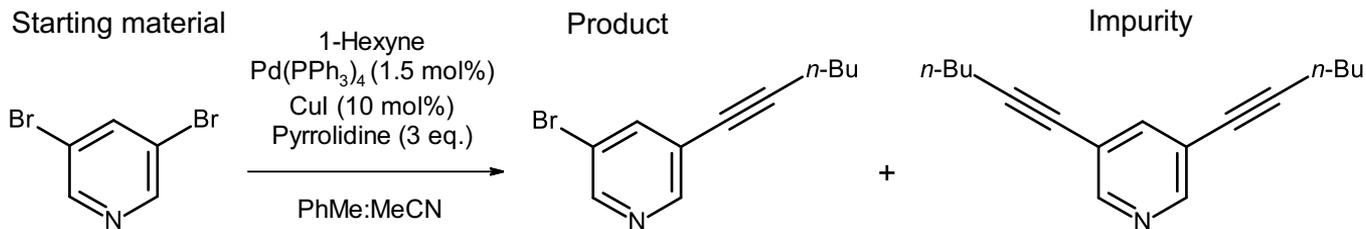
- Does optimising product yield always give the most appropriate process for the business?
- Could you sacrifice some product yield for a much better throughput when manufacturing (kg/hr)?
- Can the process only handle a certain percent of an impurity formed?



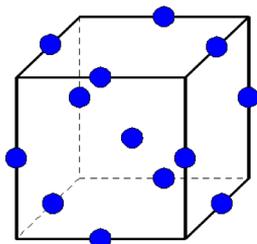
- **Machine learning multi-objective algorithm (TSEMO).**
- **Pareto front solution - One objective cannot be improved without having a detrimental effect on the other .**
- **Pareto front reveals true trade off between conflicting performance criteria.**



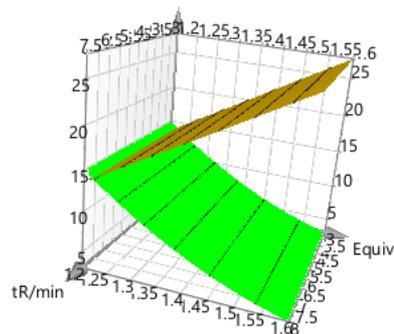
SOFR Case Study 2- Machine Learning for Process Optimisation



DOE



Response Surface Plot - E18_007507 Models (1) (MLR)



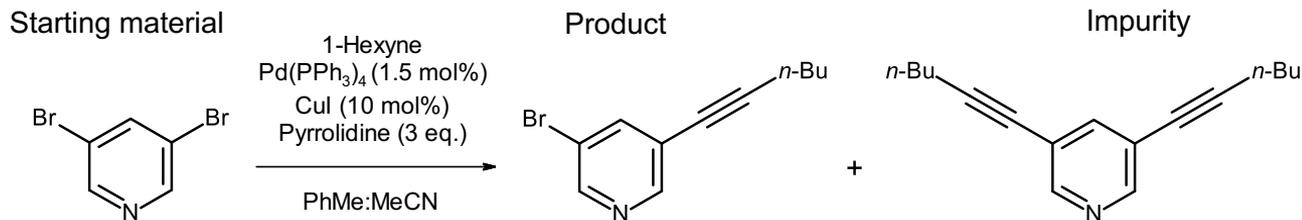
■ SM/%
■ Bis/%

DOE results

- Increased molar equivalence of Hexyne improves reaction conversion.
- However, increased Hexyne is driving impurity synthesis rather than product
- Trade off between conversion of starting material and product yield?

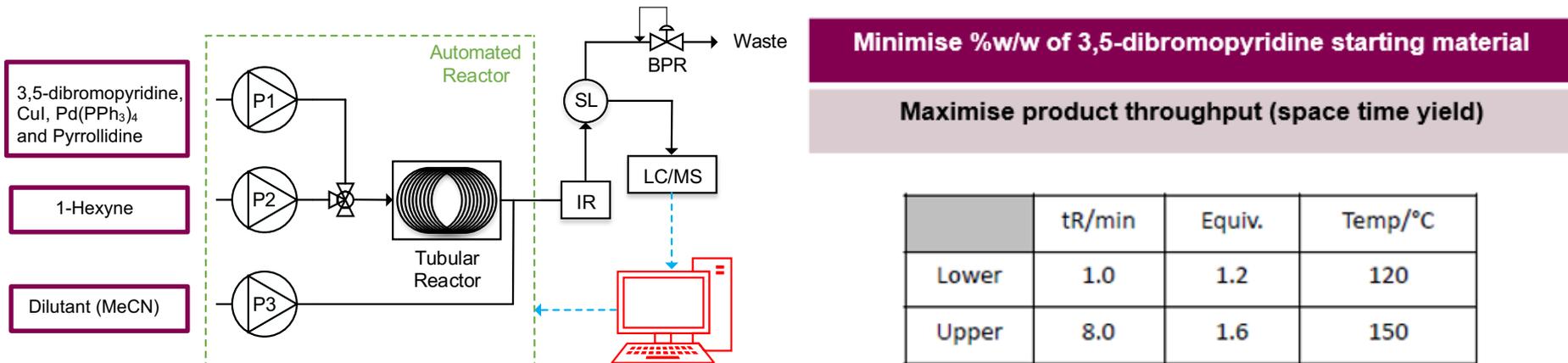


SOFR Case Study 2- Machine Learning for Process Optimisation

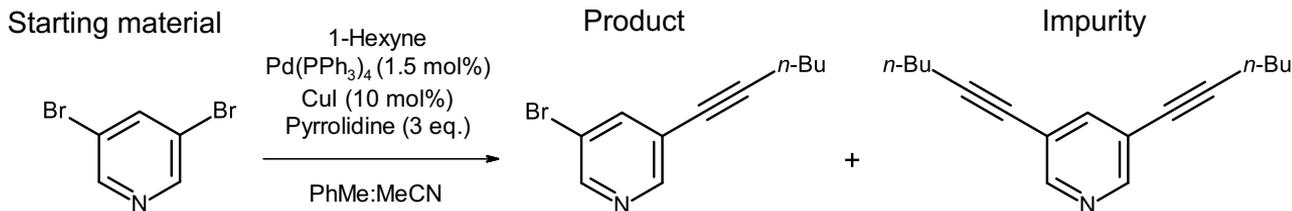


- Leeds University has shown potential of self-learning pareto multi-objective optimisation algorithm in a recent publication.

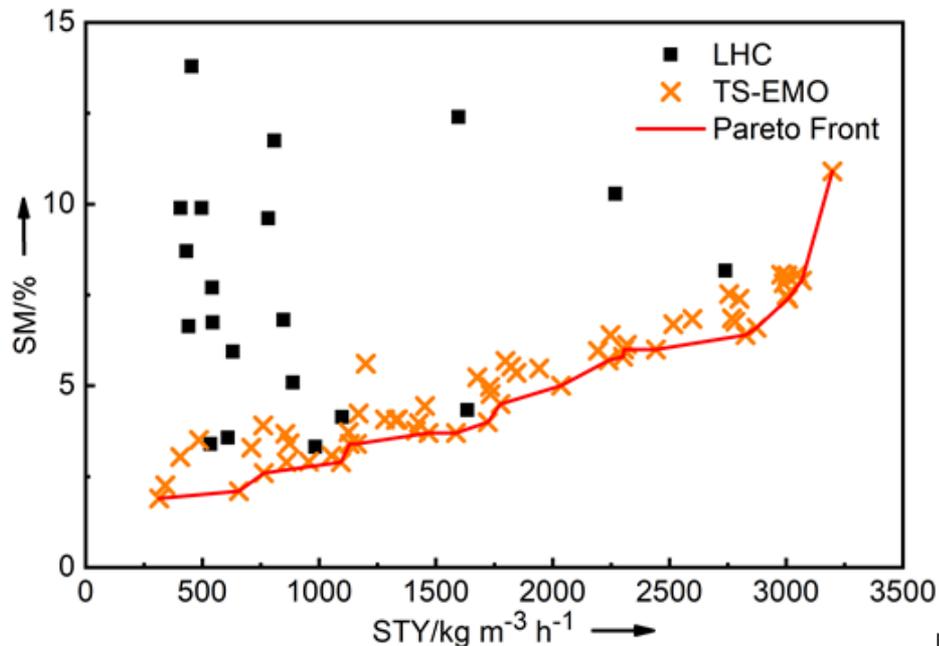
Aim – Use algorithm to assess supposed trade off between product throughput and reaction conversion suggested from DOE work.



SOFR Case Study 2- Machine Learning for Process Optimisation



- 80 reactions completed over ~35 hours
- Clear trade off between product throughput and reaction conversion
- Optimal SM %w/w being 1.9%
- Optimal space time yield 3200 kg m⁻³ h⁻¹



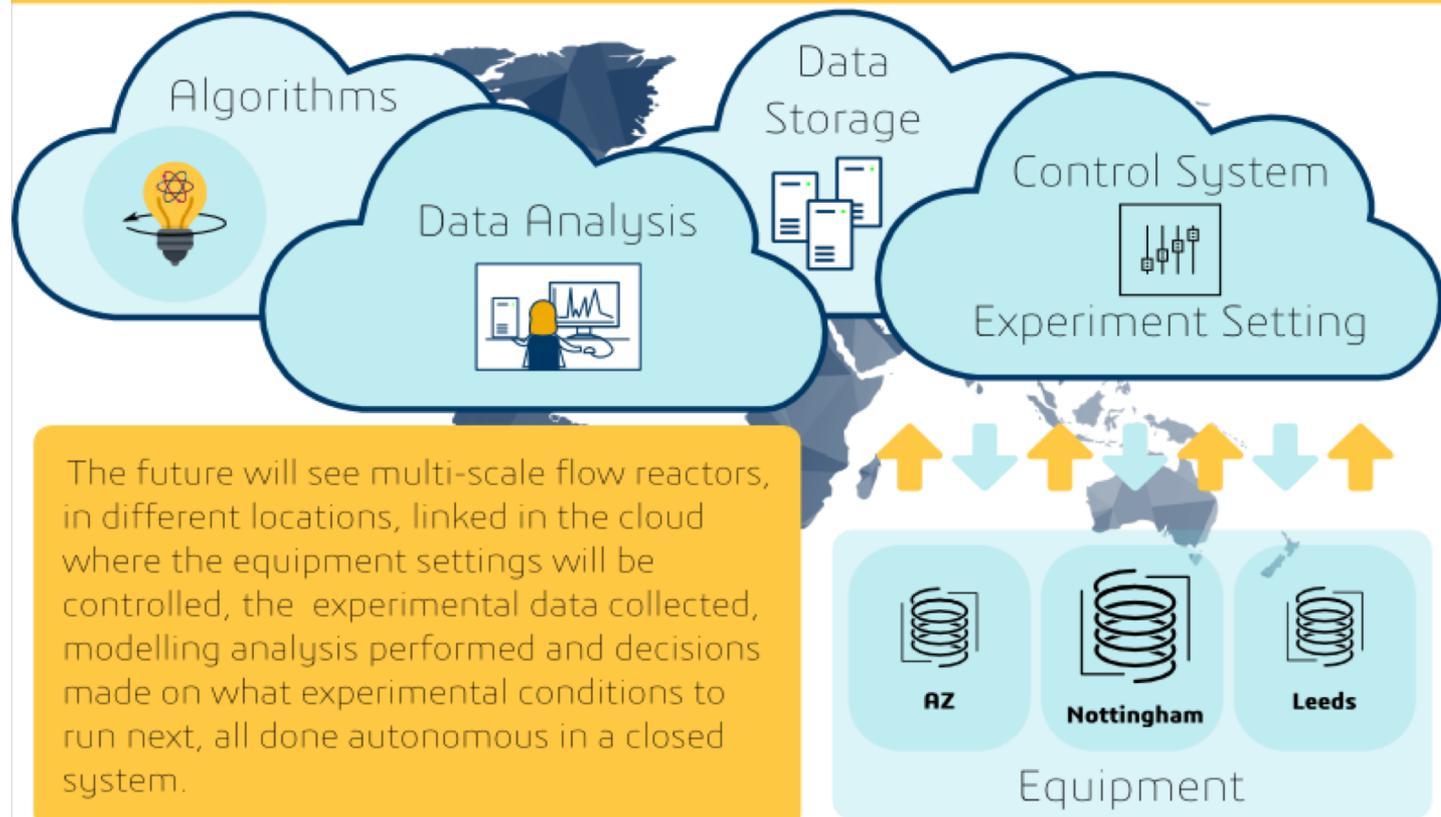
The Future – Automation for Efficient Manufacturing

Cognitive Chemical Manufacturing

- £2.5M Project using Machine Learning to Optimise Chemical Manufacturing starting July-Sept 2018
- EPSRC call on ‘Digital Manufacturing Potential’
- Led by IPRD Leeds (PI: Richard Bourne)
- Academic Partners: University College London, University of Nottingham, Hartree Centre
- Industrial Support: IBM, AstraZeneca, Swagelok and Promethean Particles
- 4 year project with 14 years of PDRA time



Cognitive Chemistry Manufacturing



Acknowledgements



- Anne O’Kearney McMullan
- Mubina Mohammed
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- Niall McCreanor
- Richard Hart
- Richard Bourne
- Adam Clayton

Thank You!



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