

Analytical Challenges and Learnings from Developing a Multi-stage Flow Process

A Chemists View

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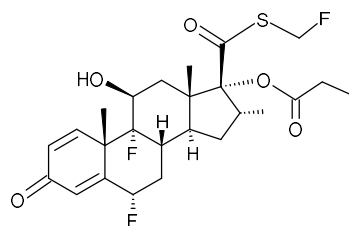


Background

- ▶ I joined the pharmaceutical industry some 37 years ago and through a series of moves and mergers ended up in process development.
- ▶ From 2000 I was mainly involved with kinetic analysis, process modelling and the development of continuous processes.

Introduction

- ▶ The activities described relate to the development of a flow process to Fluticasone Propionate (Flonase) however the challenges would be similar for any process redevelopment of an established product.



Relevant Analytical Background

- ▶ Fluticasone Propionate (Flonase) was patented in 1980 and approved for use in 1990.
- ▶ Analytical specifications reflect the available technology of the time.
- ▶ Analytical methods for isolated intermediates in the synthesis were fit for purpose for controlling quality.
- ▶ Various in house physical property specifications were developed to assess batch suitability for different formulations

Relevant Analytical Background

- ▶ The Drug substance release chromatography method had a 60 minute run time and was capable of resolving around 30 impurities.
- ▶ The methods for the intermediates were prone to artefacts if samples were left for some time.
- ▶ From a reaction monitoring and kinetic analysis perspective these methods left a lot to be desired as reaction samples would require multi-hour stability. However the drug substance method did prove valuable in tracking impurities through the process.

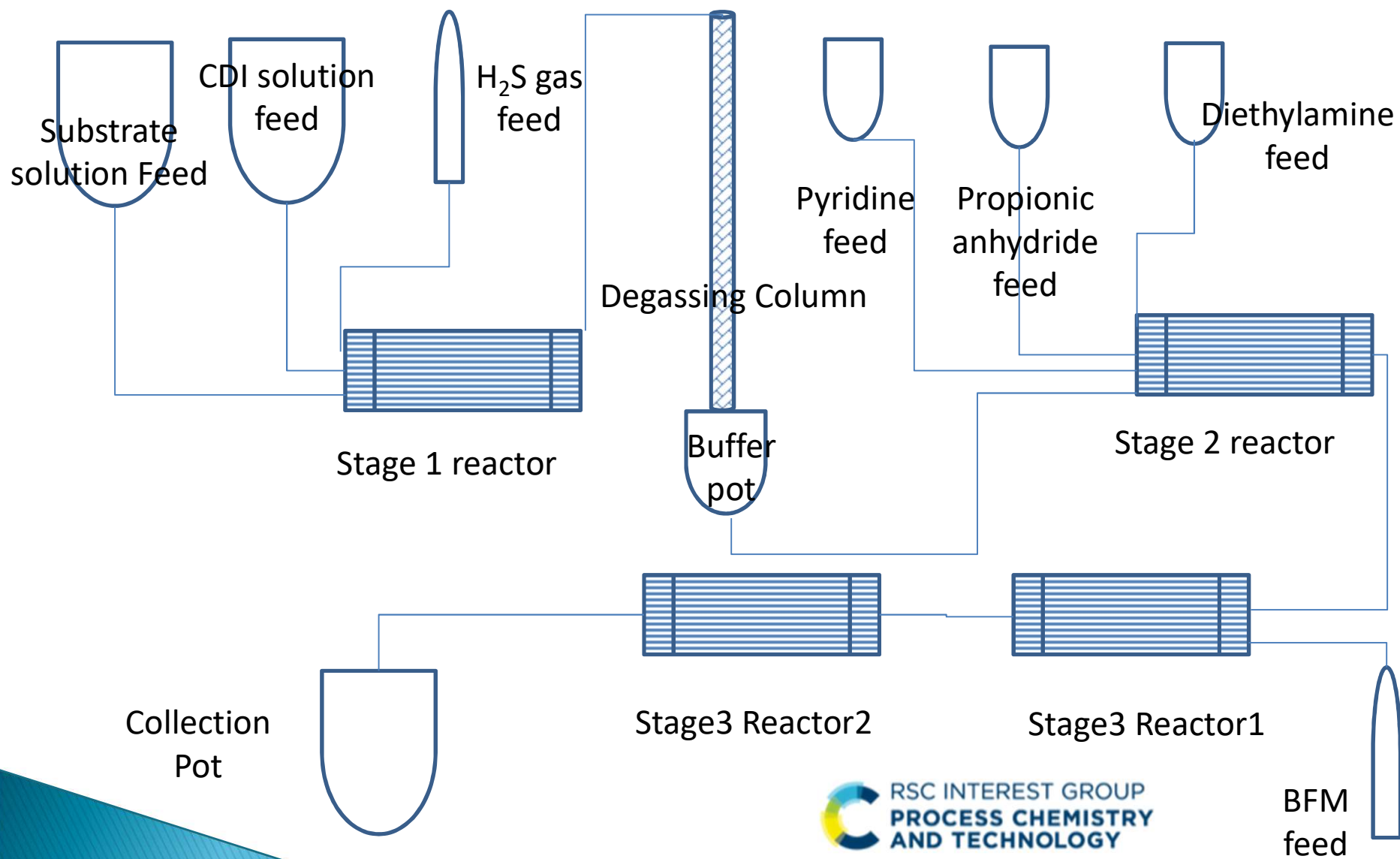
Challenges Due to flow Chemistry

- ▶ Unless the reactions are very fast an understanding of the reaction kinetics is essential in a flow process so a quantitative profile of the reaction vs time is required.
- ▶ The determination of solution concentrations in a dynamic system is significantly more challenging than end point determination.
- ▶ Typically a quench to generate a stable solution from which the reaction concentrations can be inferred is used.

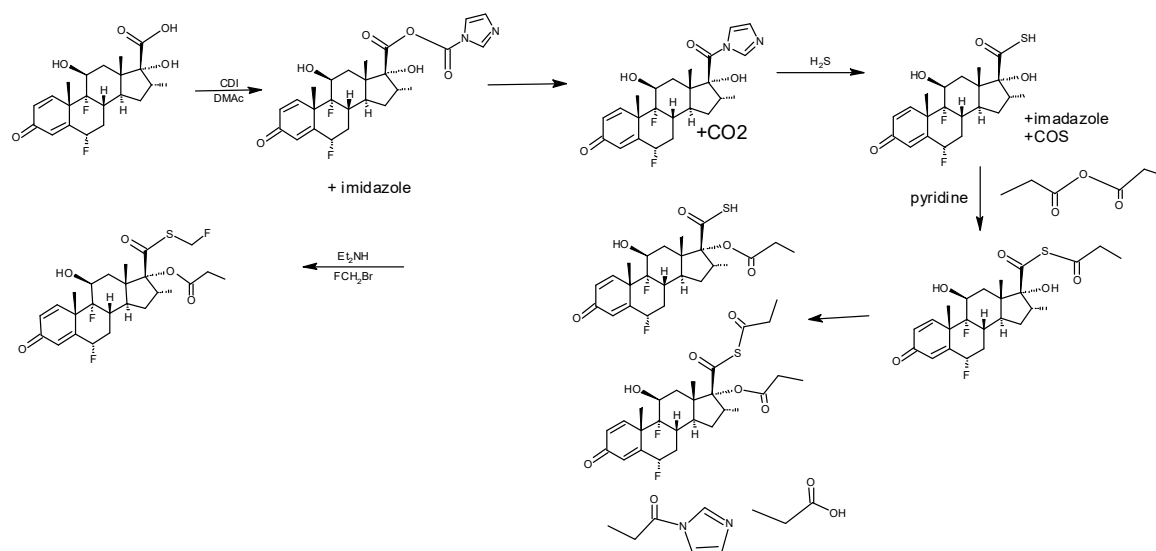
Challenges Due to flow Chemistry

- ▶ Simple dilution (after all 10ul into 1ml typical sample prep is a 100fold dilution) may be adequate to slow the chemistry sufficiently for offline analysis assuming any Intermediates are stable to the analytical conditions
- ▶ In situ analysis with various analytical probes (eg ir, uv, conductivity, nmr) can yield valuable data but benefit from offline separation techniques to aid calibration and impurity profiling.

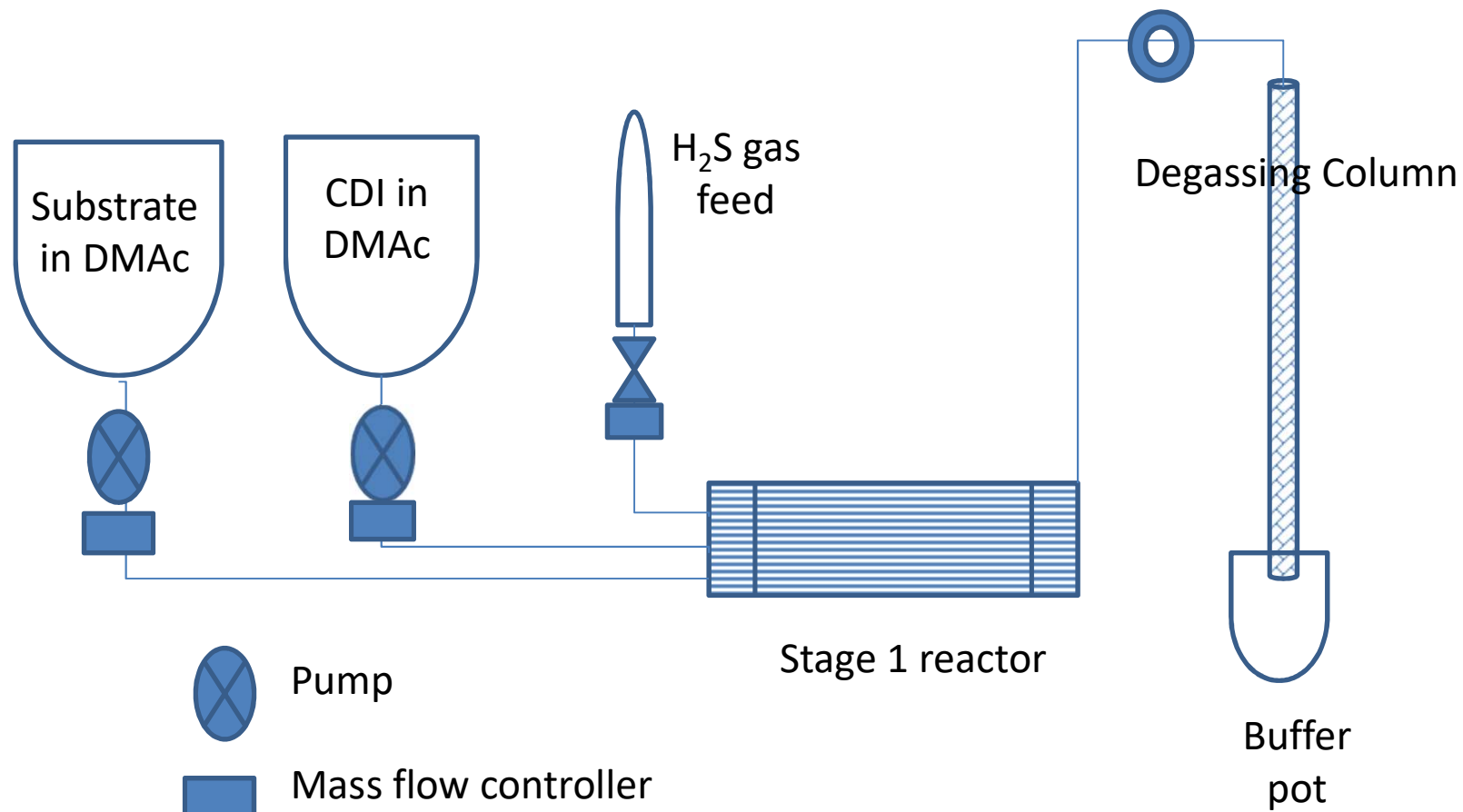
Process Schematic







Stage 1 Chemistry

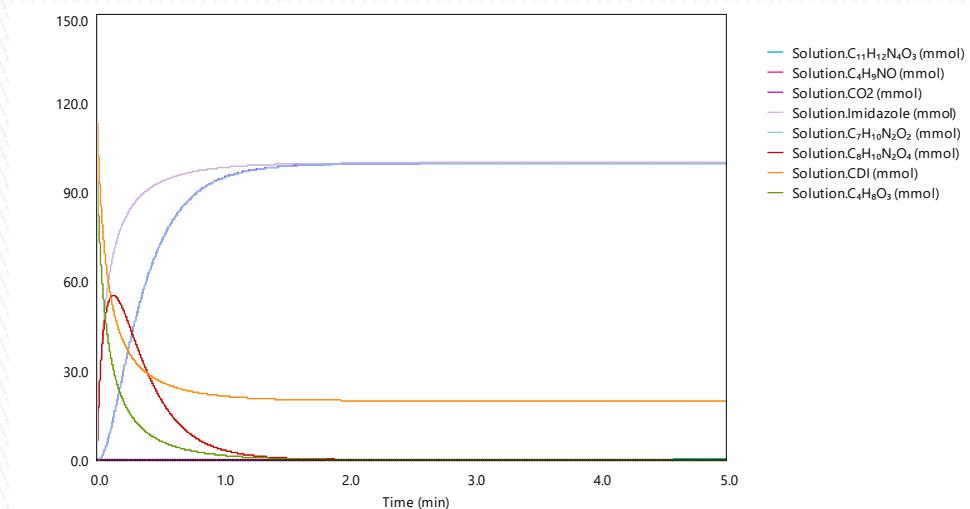


Process Schematic Stage 1



-  Pump
-  Mass flow controller
-  Control valve
-  Equilibar pressure regulator

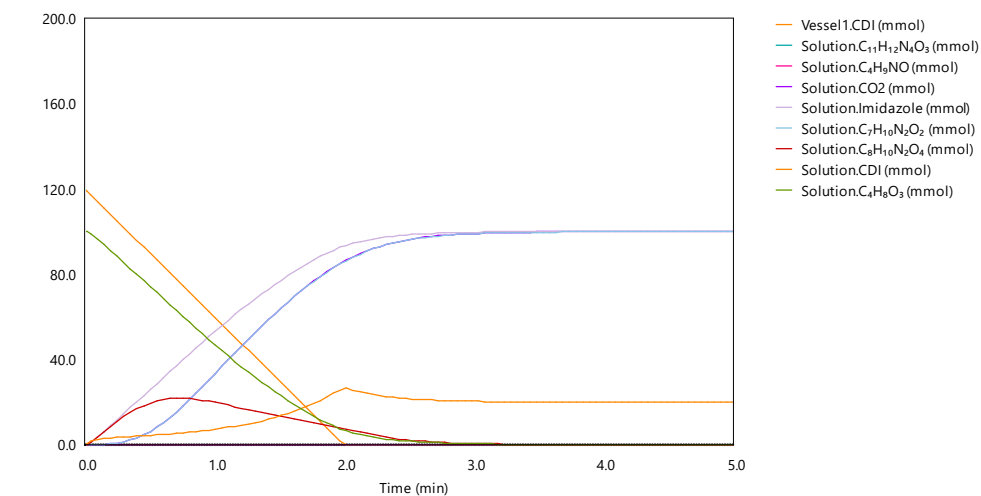
CDI Reaction Profile



Plug flow Reactor

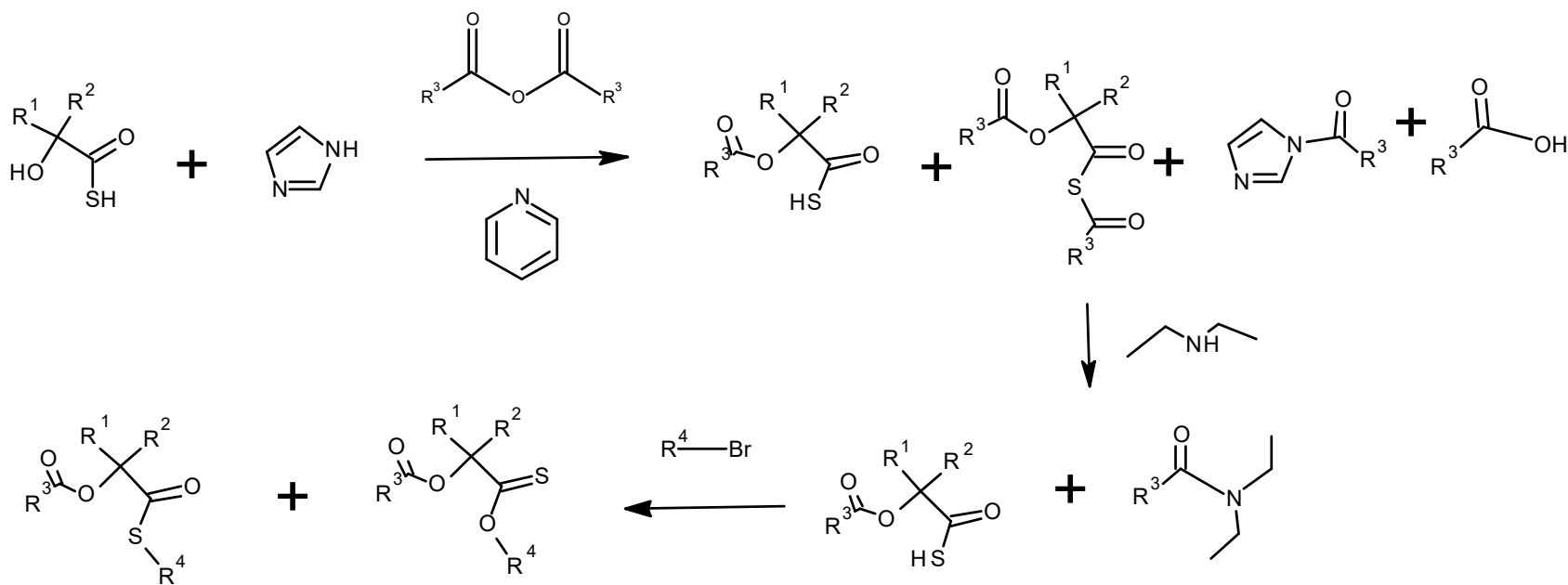


Var 1 of Base Case

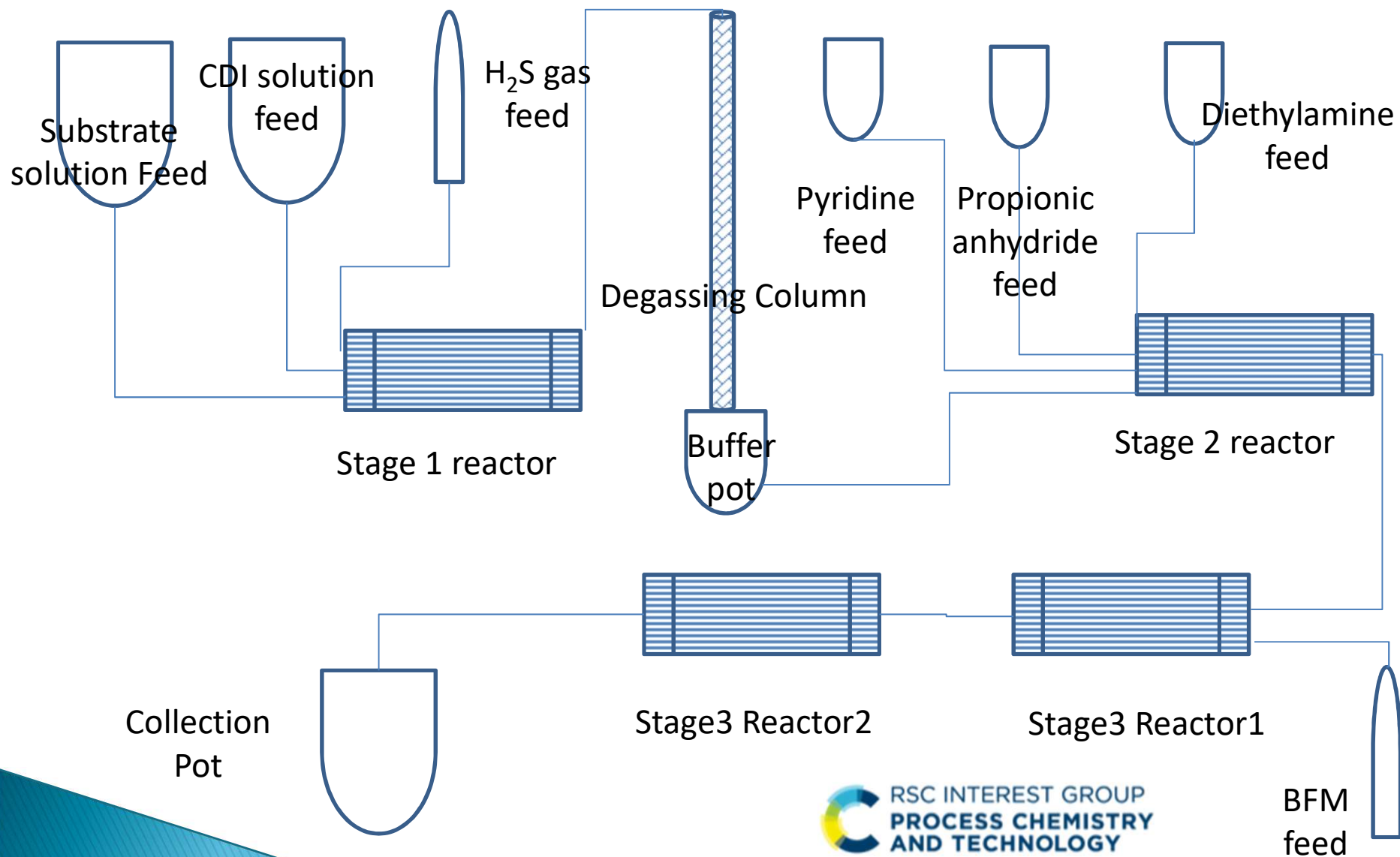


Batch Reactor
2minute Dosing

Stages 2 and 3



Process Schematic



Telescoped Processes

For

- Reduced number of vessels used.
- Reasonable process velocity.

Against

- ▶ Complex from process instructions and operator point of view.
- ▶ Increased risk of maloperation.
- ▶ Probably require in process checks.
- ▶ Cause of process failure difficult to identify
- ▶ Need cleaning verification between batches

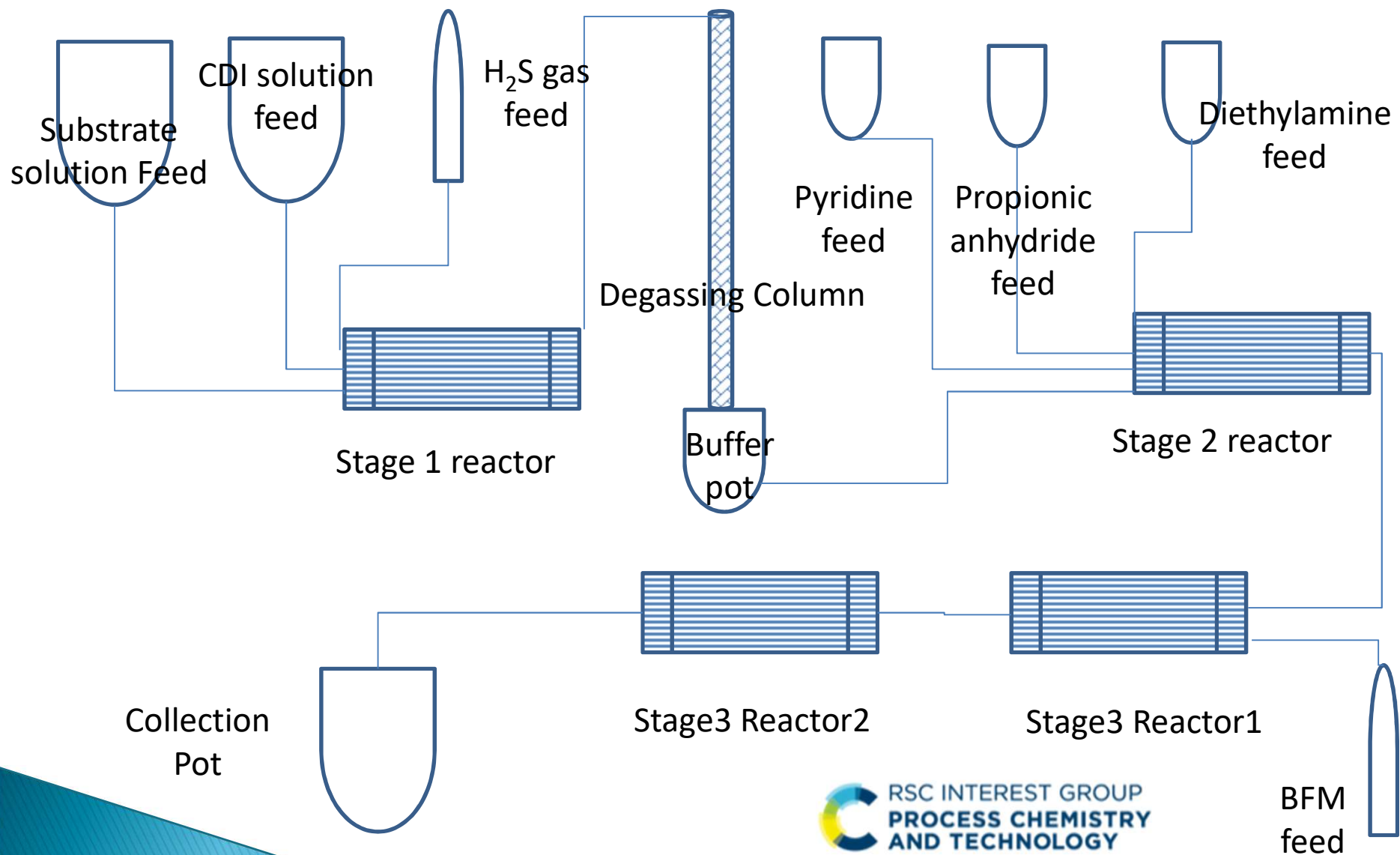
Telescoped vs multistage flow

- ▶ Able to run in process checks and take corrective action eg stir for longer add more reagent prior to moving to next step.
- ▶ Process can be monitored continuously but unable to correct material that has already moved on. Each step is segregated stage 1 reactor never sees stage 3 so less risk of contamination

Telescoped

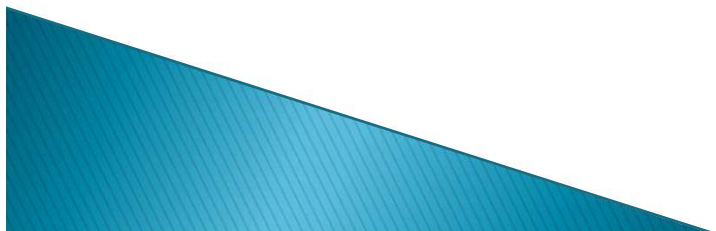
Multi stage flow

Process Schematic




Process Optimisation and Design Space Definition

- ▶ Both multistage flow processes and telescoped batch processes suffer from too many variables.
- ▶ This process has 9 flow rates 4 reactor temperature, 2 solution feed concentrations and a buffer pot fill level 16 parameters in all.
- ▶ A large part of this must be reduced by process understanding as the experimental demand is just too large to explore the whole design surface.

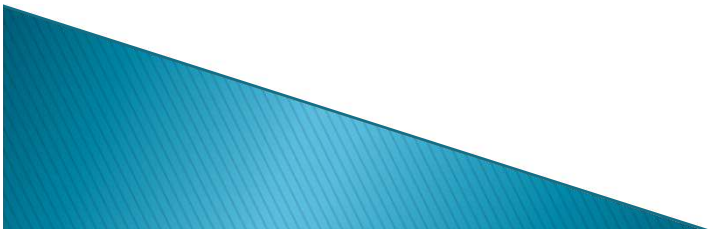


Summary

- ▶ Avoiding solids will make your process easier but be prepared to see them at unexpected moments.
 - ▶ In developing a flow process you will have to change your mindset on what constitutes an independent variable.
 - ▶ A good understanding of the reaction kinetics and impurity profile over the whole process is essential.
 - ▶ Incomplete earlier steps are going to be your main route to failure. Design with this in mind! Ensure your analytical methods can cope with this.
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Acknowledgements

- ▶ All my colleagues “Chemists, Analysts and Engineers” over the years who delivered this process.
- ▶ Scale up systems whose software and support underpinned the kinetic understanding of this process



Q and A

